Charlotte-Rock Hill-Gastonia MSA Summertime Ozone Formation





Photos from 2001 Mecklenburg Country State of the Environment Report

Arizona's Valley of the Sun "Brown Cloud"



Phoenix photo from http://phoenix.about.com/library/weekly/uc051601a.htm, accessed Mar 2004

- Observed since early 1990's; contains C Particles and NO₂ (brown) gas
- From burning fossil fuels: cars, construction equipment, power plants, lawn mowers, leaf blowers contribute to this brown cloud
- Weather also a key factor

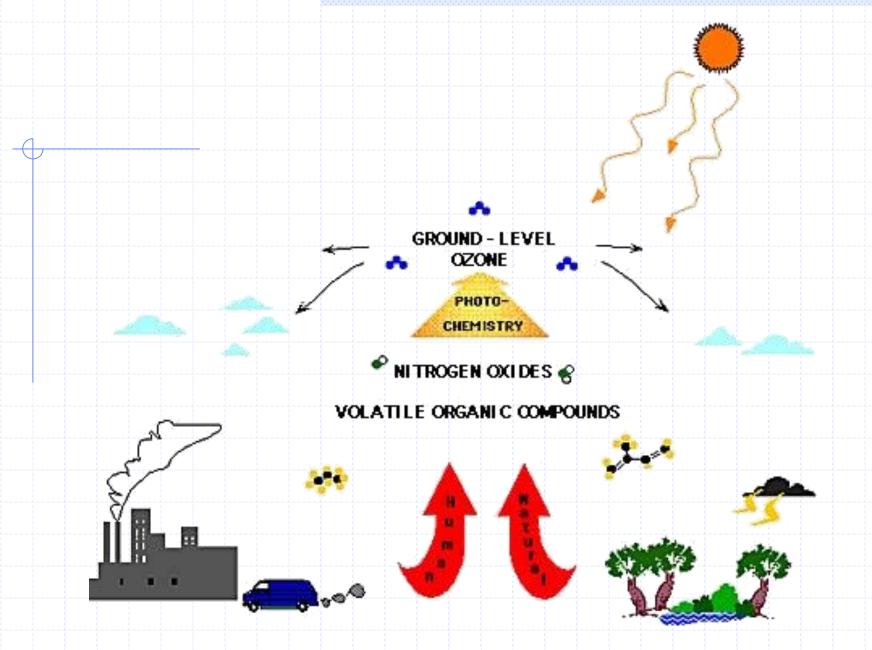


Image take from http://www.al.noaa.gov/WWWHD/Pubdocs/TropoRural.html, accessed Mar 2004

Presence of NO₂ (brown gas) results in inevitable formation of ozone

NO₂ (brown) + sunlite
$$\rightarrow$$
 NO• + O•
$$0 + O_2 \rightarrow O_3 \text{ (Ozone)}$$

• Control measures to effectively address ground-level ozone must focus on minimizing NO₂ formation

Ozone precursors

Nitrogen Oxides

- •Dry Air is 78% Nitrogen (N₂), 21% Oxygen (O₂)
- Heating air to high temperatures forms NO. (motor vehicles, fossil-fuel power plants, lightning)
- NO• is an ozone precursor and also reacts directly with ozone to destroy it

Volatile Organic Compounds (VOC's)

- •VOC's continuously emitted by fuel spills, motor vehicles, trees, air fresheners, paints, and countless other sources
- •OH• hydroxyl radical is the atmosphere's "natural detergent" that cleans VOC's from the air by initiating the process that converts them into carbon dioxide and water₅

Volatile Organic Compound (VOC) Removal

Air's natural VOC cleanser produced from ozone and water

$$*O \bullet + H_2O \rightarrow 2OH \bullet$$

OH● works by transforming stable VOC molecules (R-H) into reactive radicals (R●)

$$OH \bullet + R-H \rightarrow R \bullet + H_2O$$

 Reactive organic radicals immediately combine with oxygen to form peroxide radicals

$$R \bullet + O_2 \rightarrow (R-O-O \bullet)$$

Formation of Nitrogen Dioxide

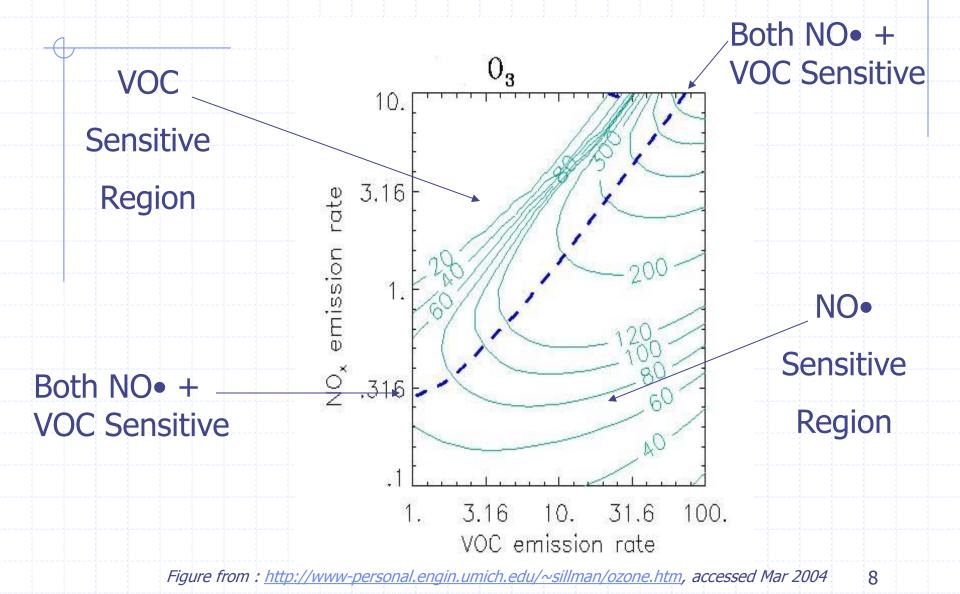
$$NO \bullet + R-O-O \bullet \rightarrow NO_2 \bullet + RO \bullet$$

Formation of NO_2 (and O_3) driven primarily by:

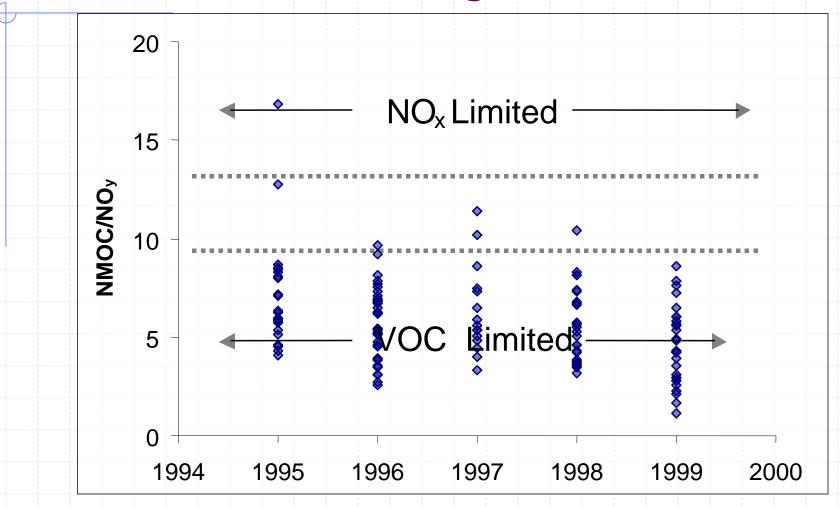
- Concentration of VOC's (near emission sources)
- Concentration of NO• (rural & downwind areas)
- Concentrations of both NO• and VOC's

***Summertime ozone pollution results from an overload in our air of NO•, VOC's, or both

NO and VOC Effects on Ozone Production

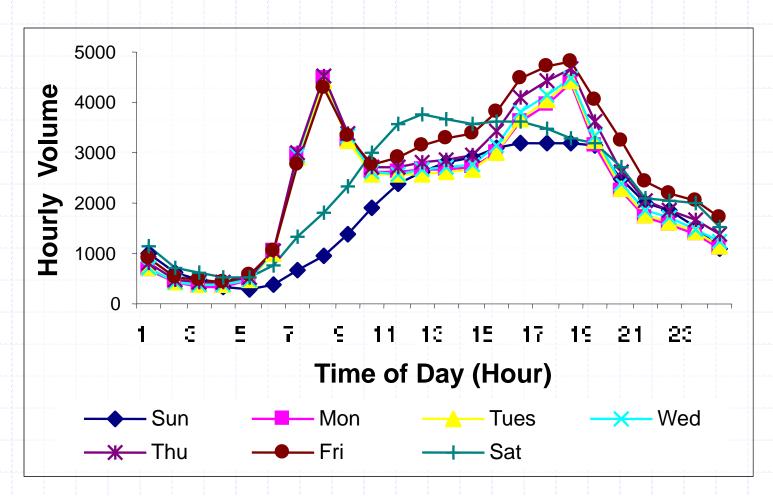


'95-'99 AM NMOC/NO_y Ratios for the Plaza Monitoring Site

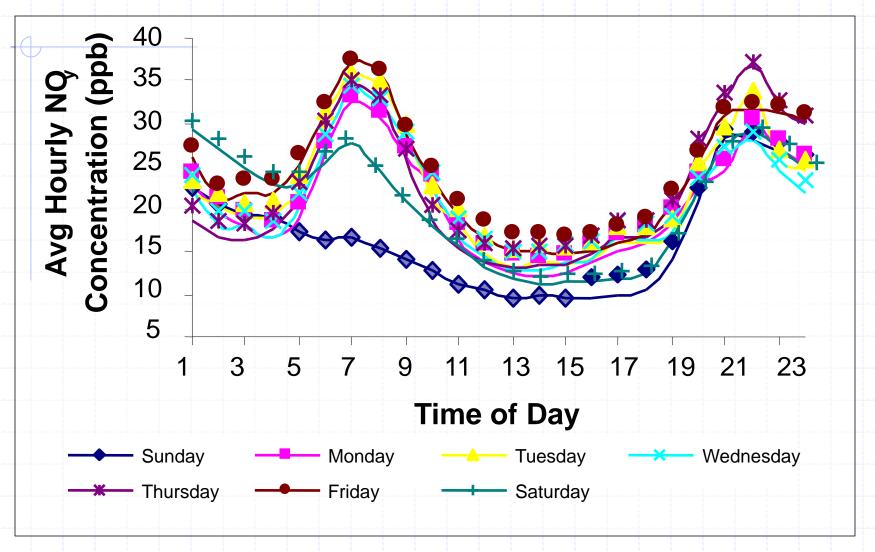


From Perry, JL and Owens, PM, "Weekday/Weekend Variability and Long-Term Trends in Traffic, CO, NO_y and

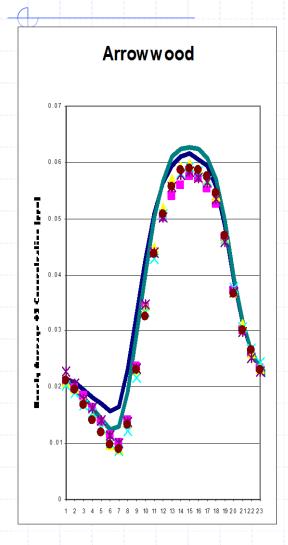
I-77 Traffic Volume vs. Time of Day

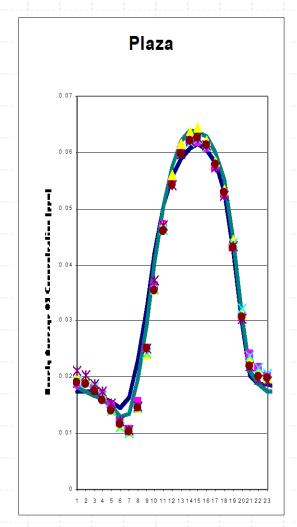


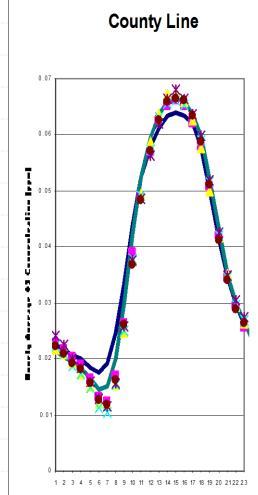
May-Sep '95-'98 Plaza Average Hourly NO_y

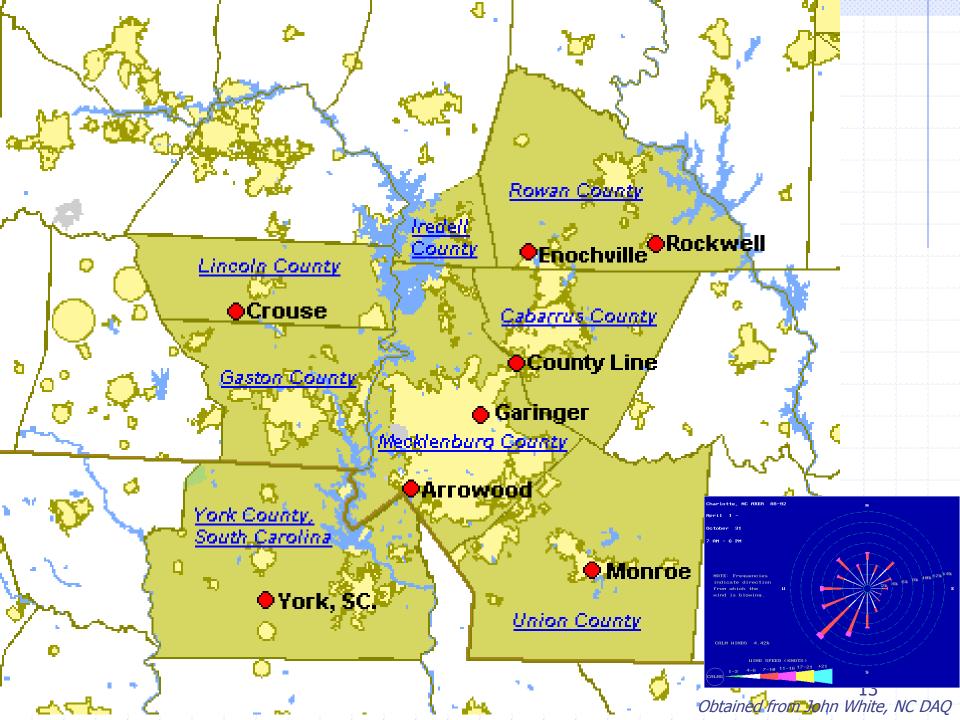


1990-1998 May-Sep Diurnal Charlotte Ozone Averages By Day of Week (*weekend days-lines, weekdays-markers*



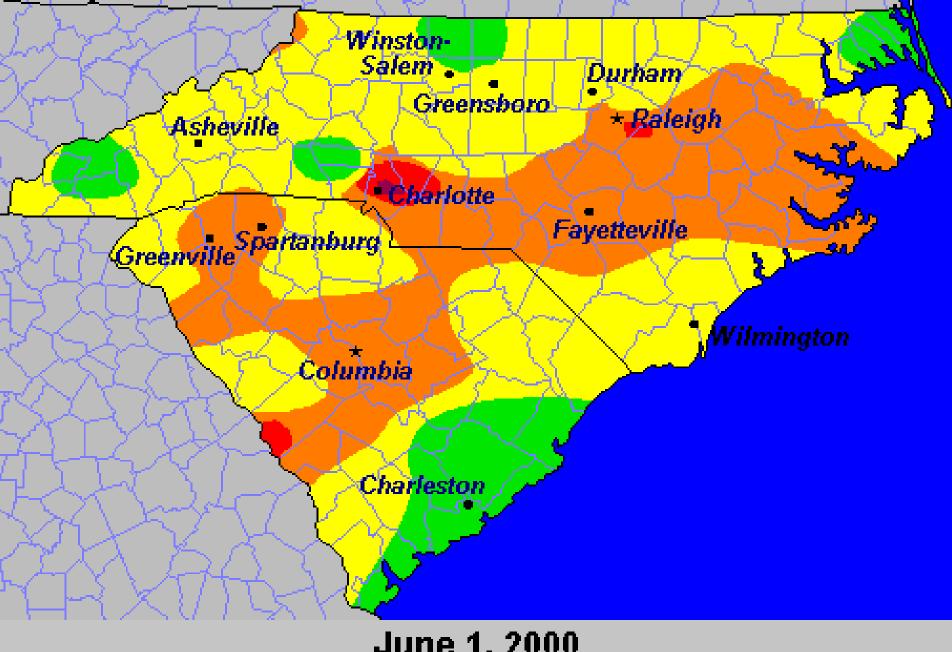








10 am June 1, 2000 EDT



June 1, 2000